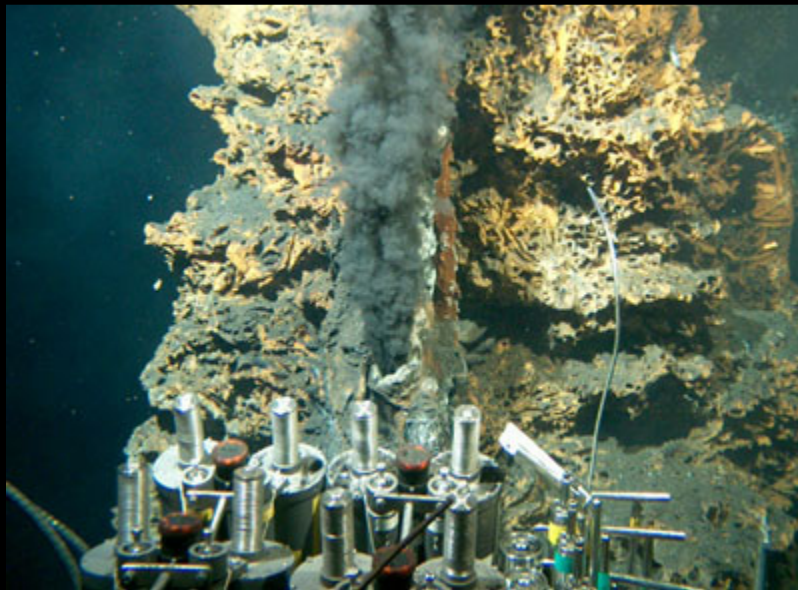


In-situ chemistry of seafloor hydrothermal vent fluids



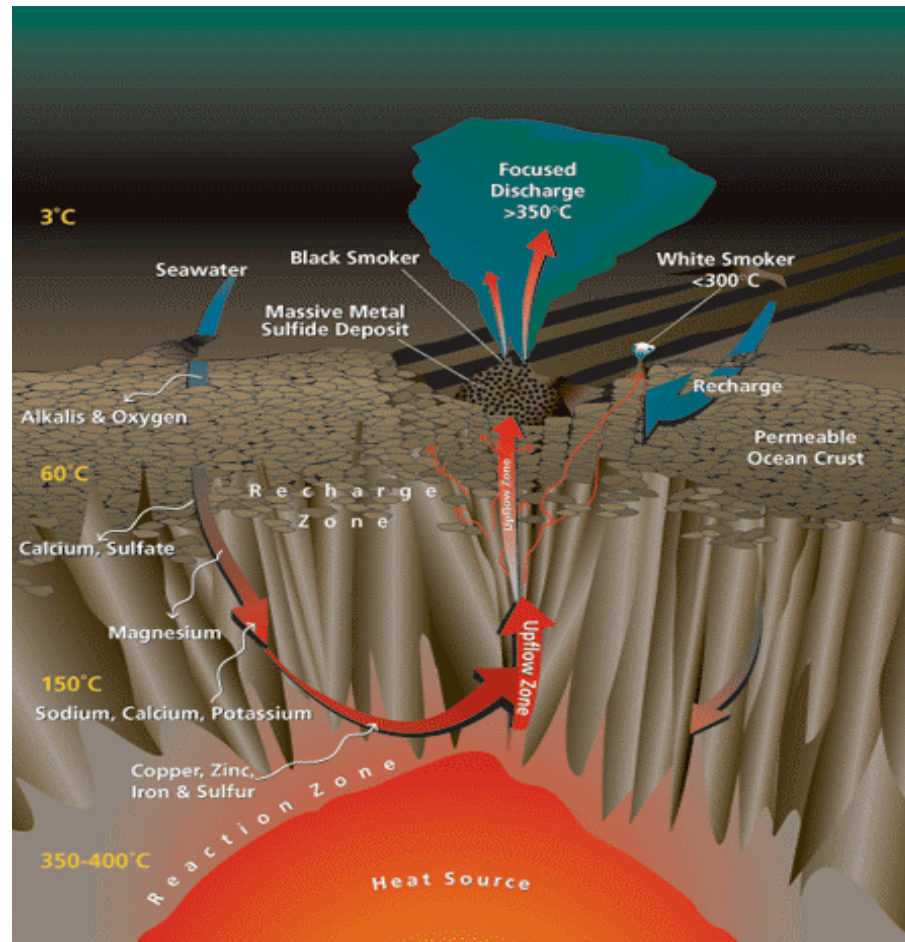
W.E. Seyfried, Jr., Kang Ding and Zhong Zhang
Department of Geology and Geophysics
University of Minnesota

Hydrothermal Processes at MOR's



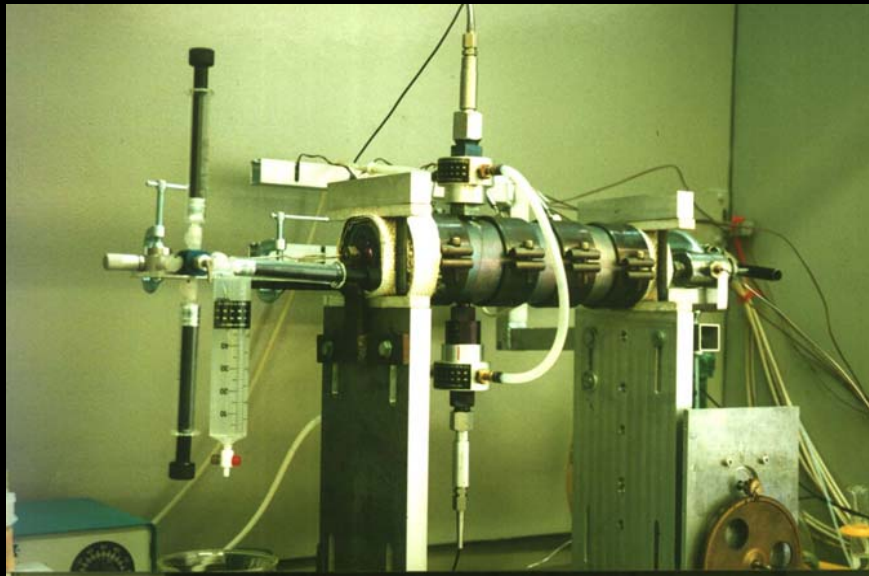
- ~25% of Earth heat loss
- Temperatures in excess of 388°C at vent
- Generally acidic measured pH
 - Non-conservative
- Moderately reducing
 - Non-conservative
- Critical need for *in-situ* measurements for pH and redox

Subseafloor Hydrothermal System



S. E. Humphris (WHOI)

Ti flow-through vessel for sensor development at high temperature and pressure condition



pH Sensor



$$pH_{T,P} = \frac{F}{2.303RT} \{ \Delta E_{\text{cell}}^{\circ} - \Delta E_{\text{meas.}} \} - \log \left(\frac{a\text{Cl}^-}{a\text{H}_2\text{O}^{1/2}} \right)$$

Dissolved H₂ Sensor



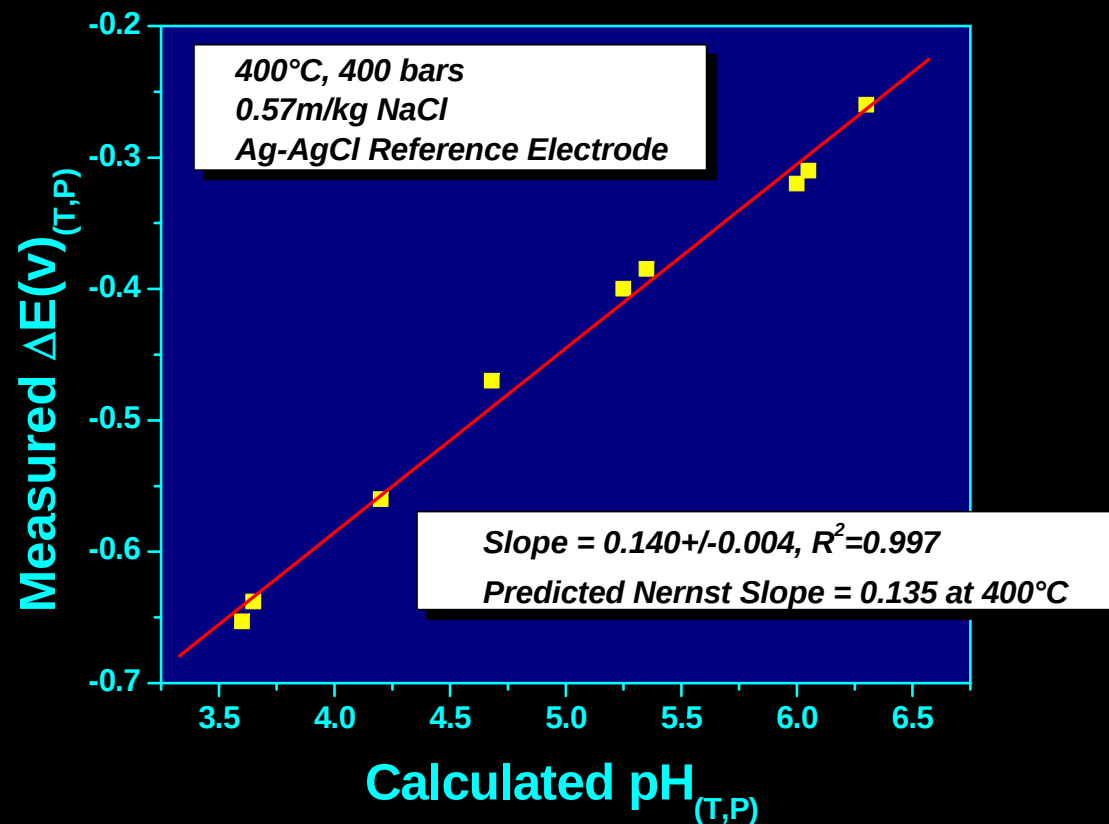
$$\log a\text{H}_2 = \frac{2F}{2.303RT} \{ \Delta E_{\text{meas.}} - E_{\text{cell}}^{\circ} \} - \log \left(\frac{K_{\text{H}_2}}{a\text{H}_2\text{O}} \right)$$

Dissolved H₂S Sensor

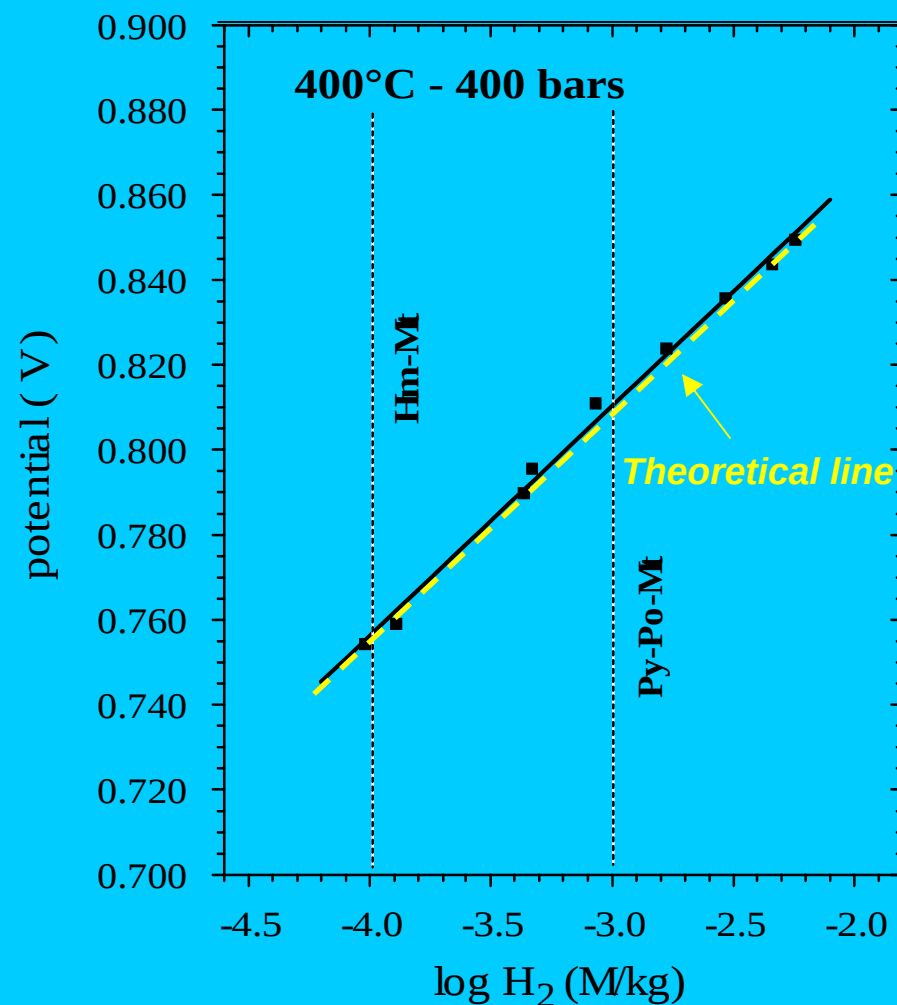


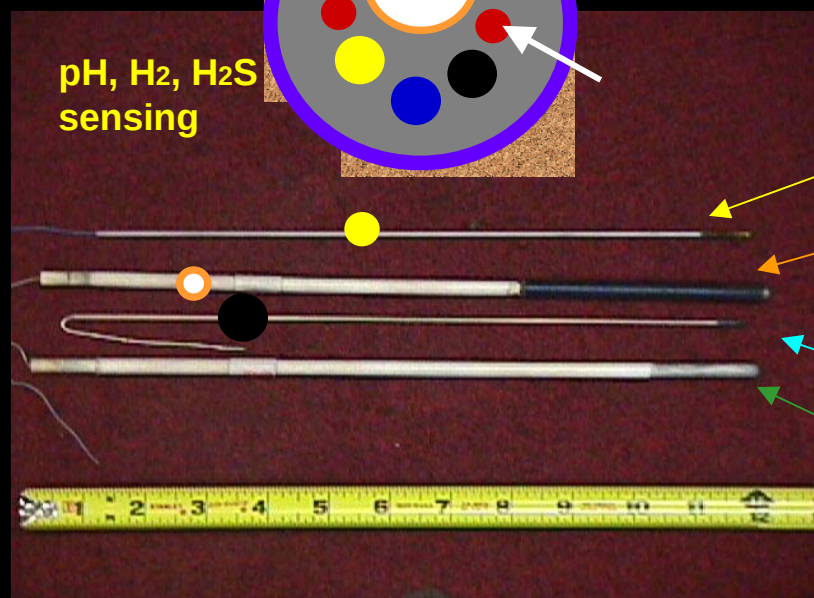
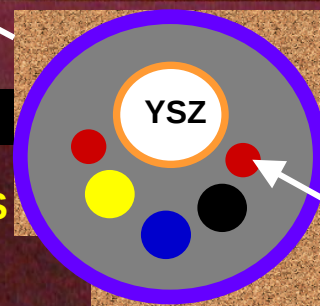
$$\log a\text{H}_2\text{S} = \frac{2F}{2.303RT} \{ \Delta E_{\text{meas.}} - E_{\text{cell}}^{\circ} \} - \log \left(\frac{K_{\text{H}_2\text{S}}}{a\text{H}_2\text{O}} \right)$$

Lab Measured Potential (YSZ) against Theoretically Predicted (HKF) pH(in-situ)



Experimental Calibration of YSZ-Au Sensor for H₂ Determination





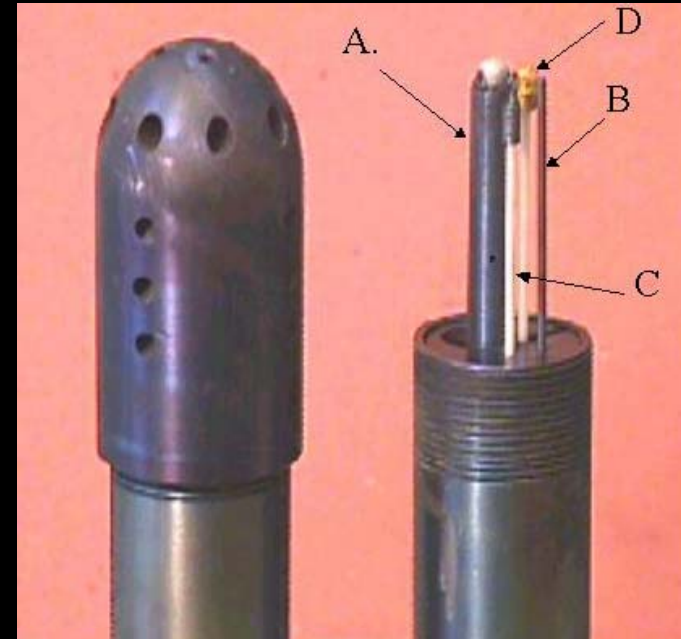
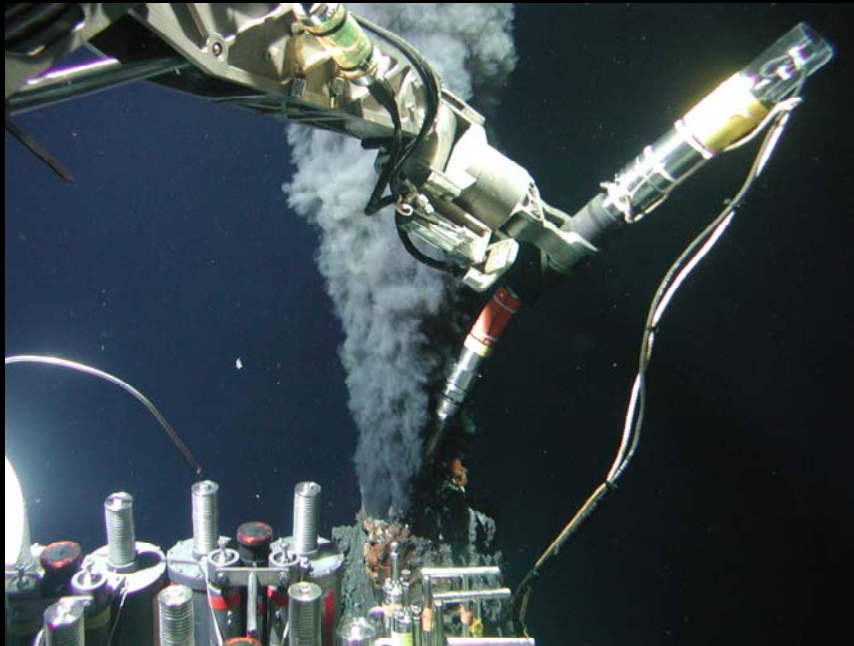
Au (H_2)

YSZ (pH)

Ag/Ag₂S
(H₂S)
Sensor

Ag/AgCl

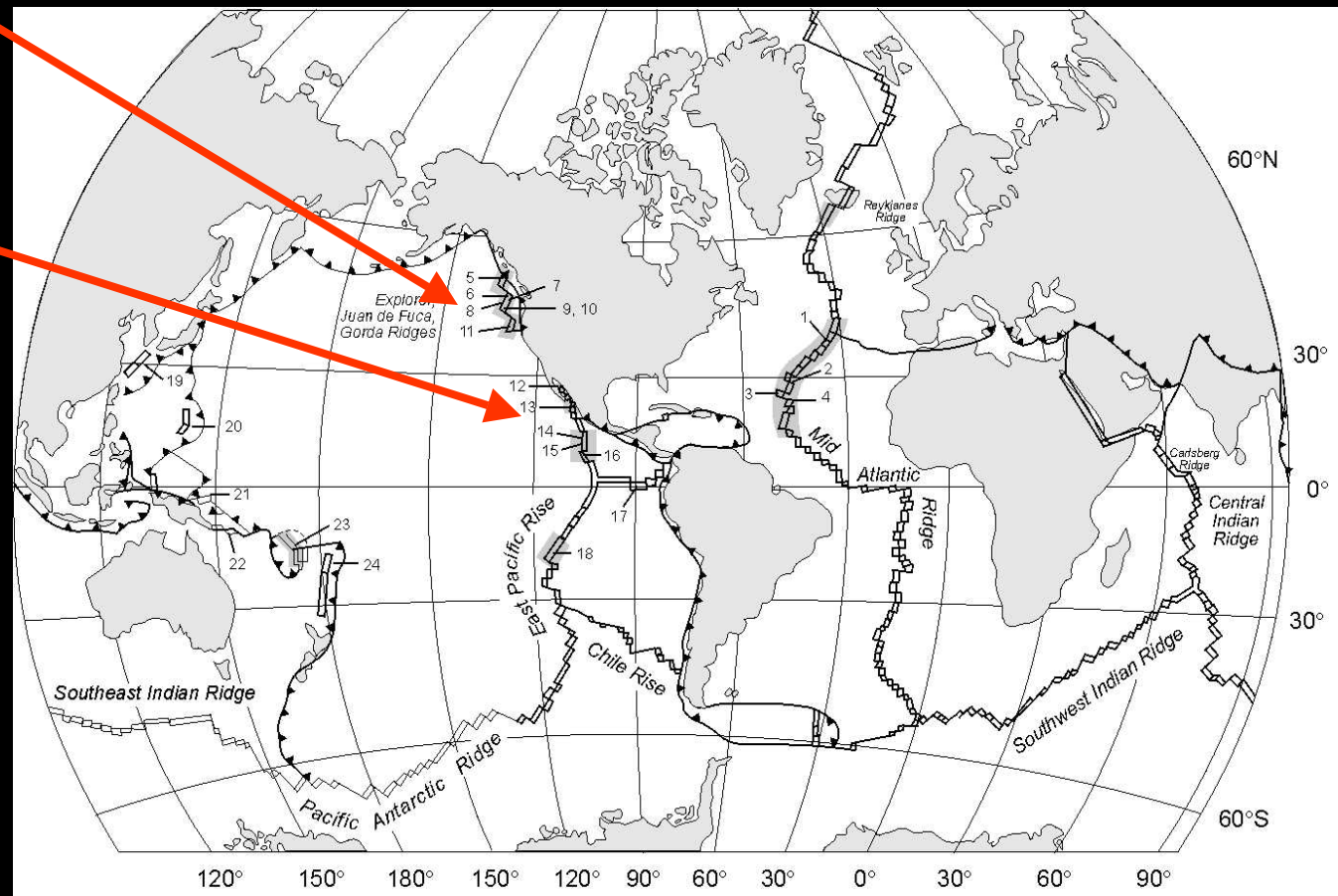
Seafloor Deployment: Real Time



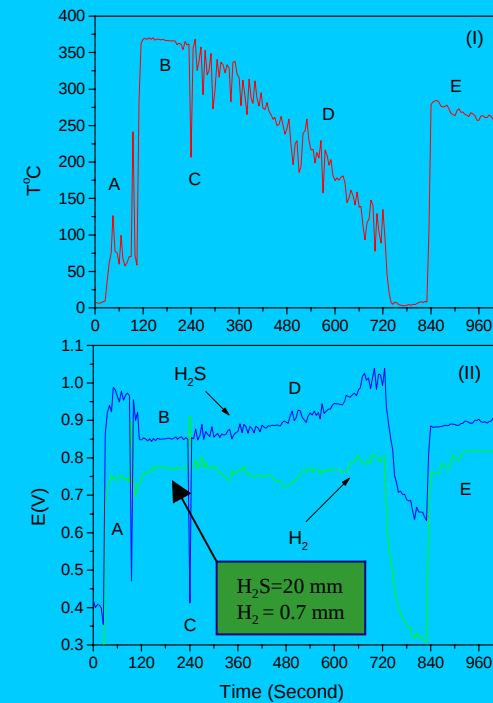
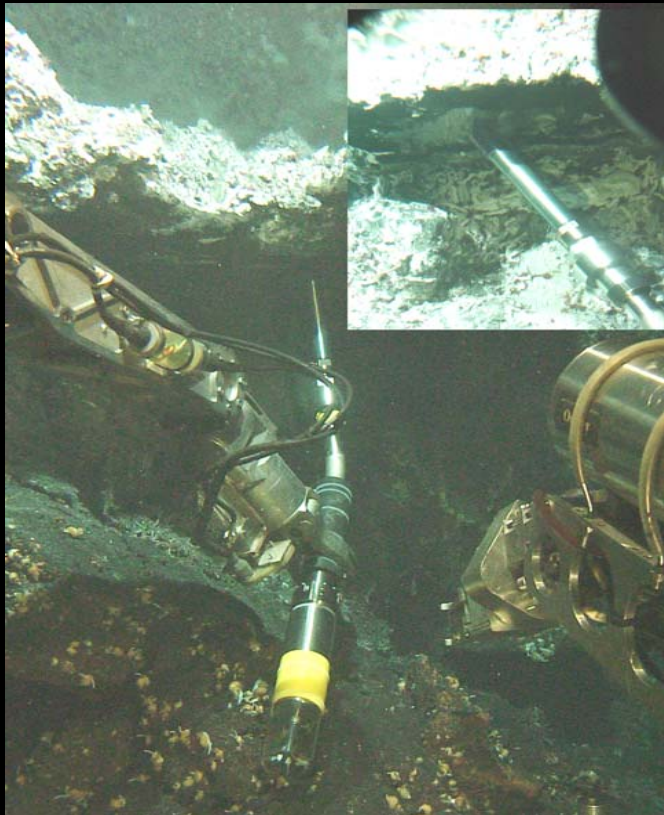
Chemical Sensor Studies of Hydrothermal Vents at Mid-Ocean Ridges

Main Endeavour
Field, JDF

EPR, 9-10°N

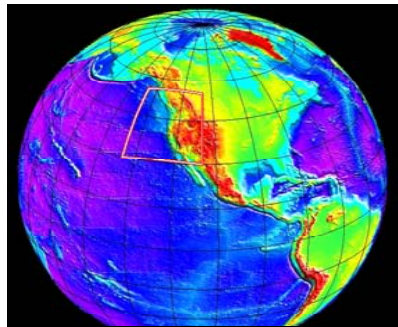


In-situ Measurements of H₂ and H₂S at MEF, JDF



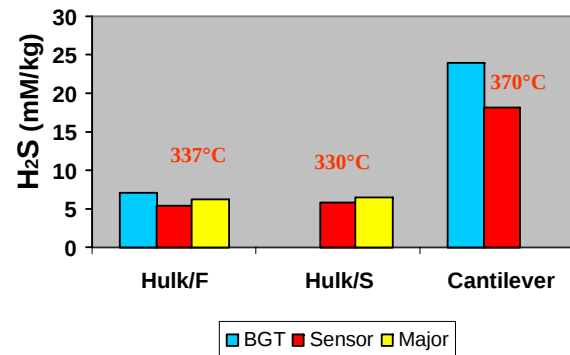
Ding et al. 2001

In-Situ Dissolved Gases at MEF- Comparisons with Conventional Measurement Techniques

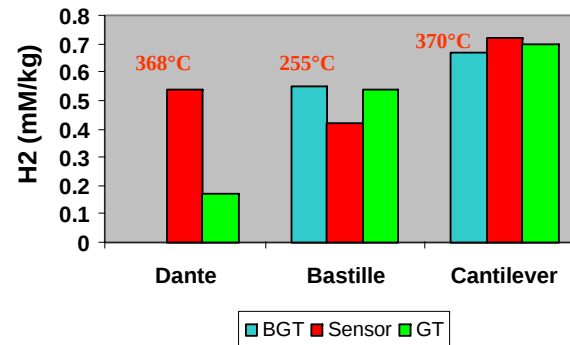


Endeavour vent field
Juan de Fuca Ridge

Comparison of dissolved H₂S concentrations measured by using different methods

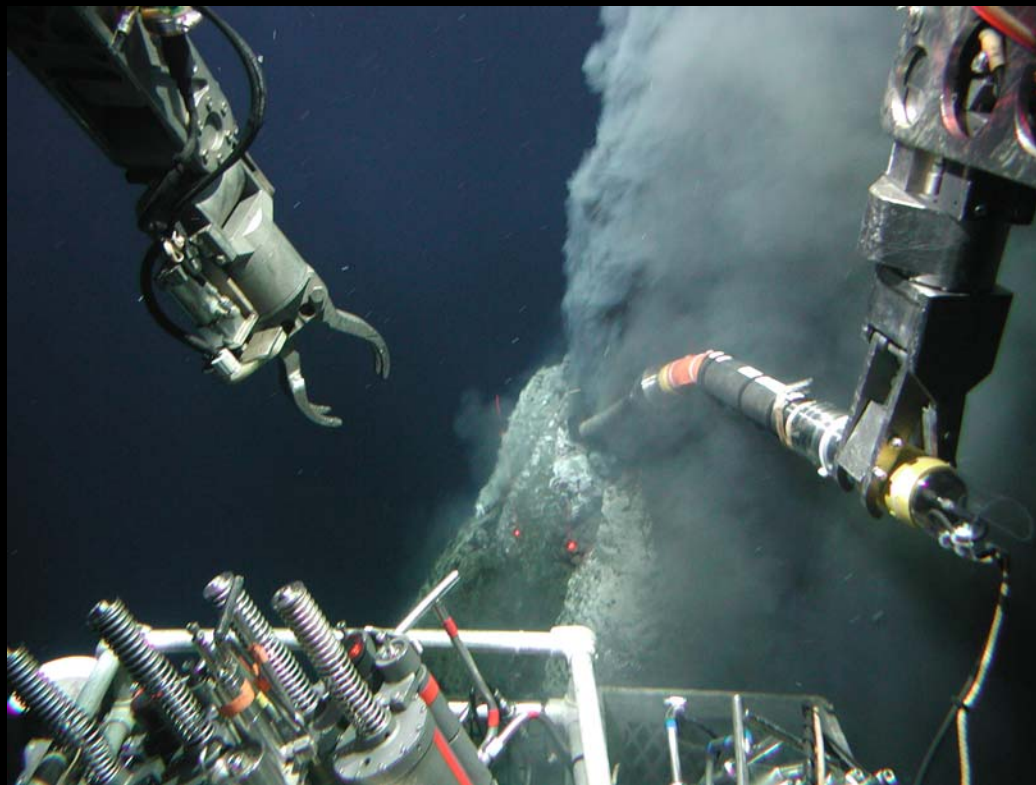


Comparison of dissolved H₂ concentrations measured by using different methods

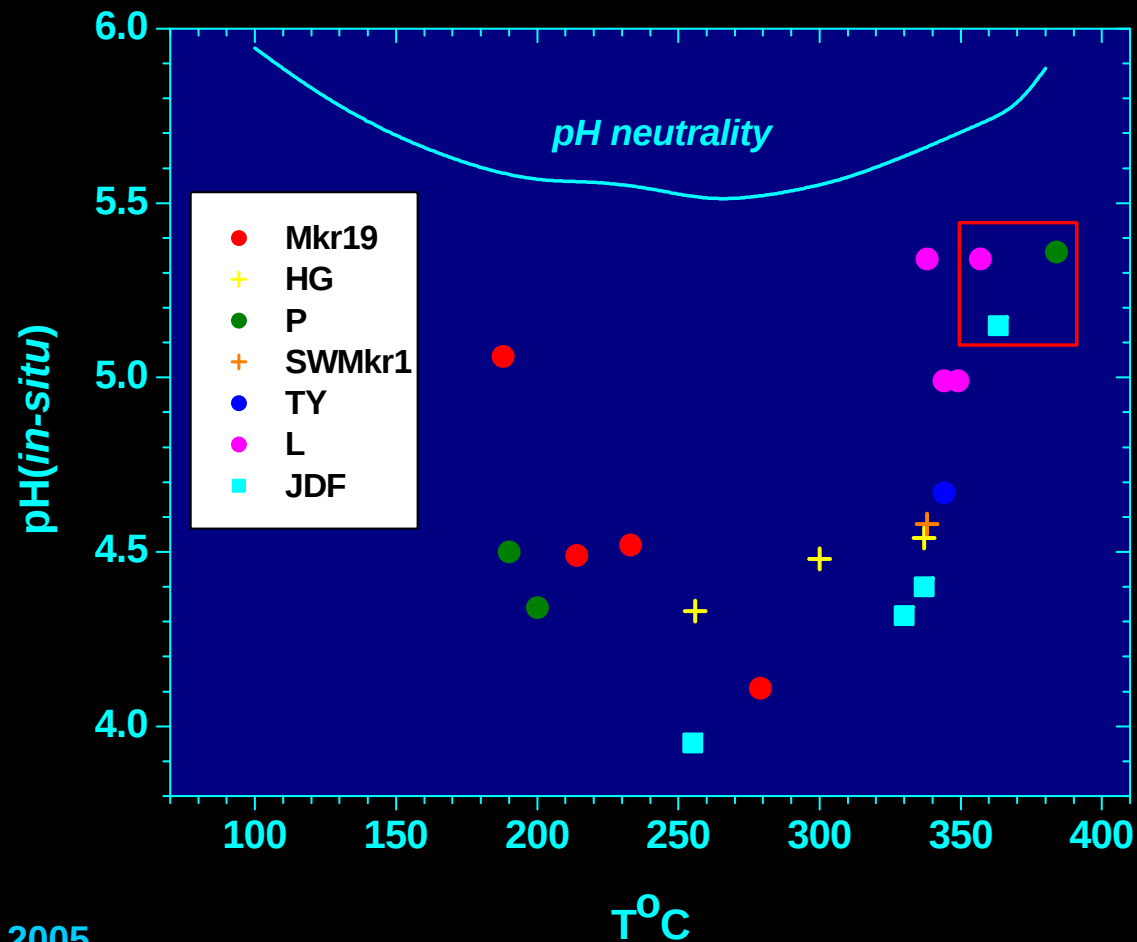


Ding et al., EPSL, 2001

In-situ Controls on Vent Fluid-Seawater Mixing

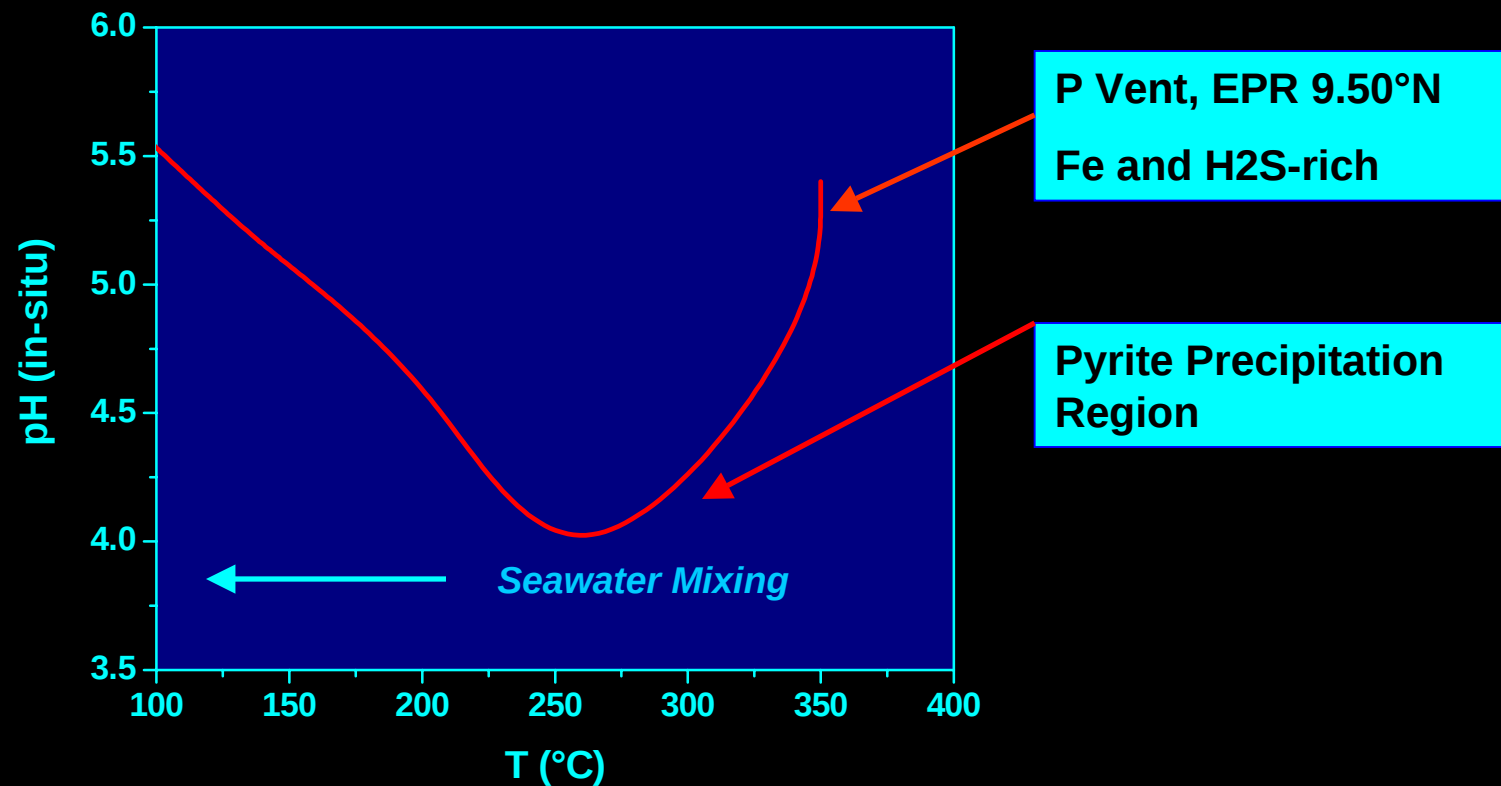


In-Situ pH versus Vent Temperature

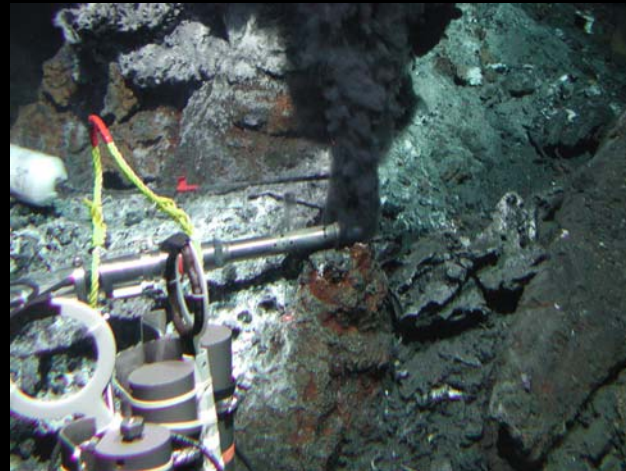
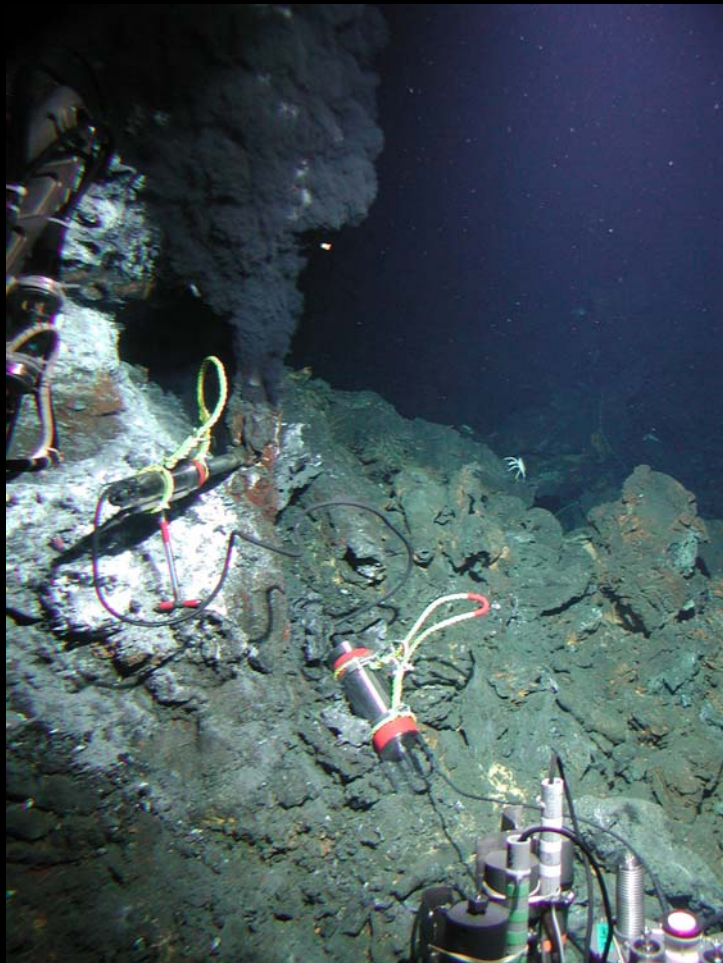


Ding et al. 2005

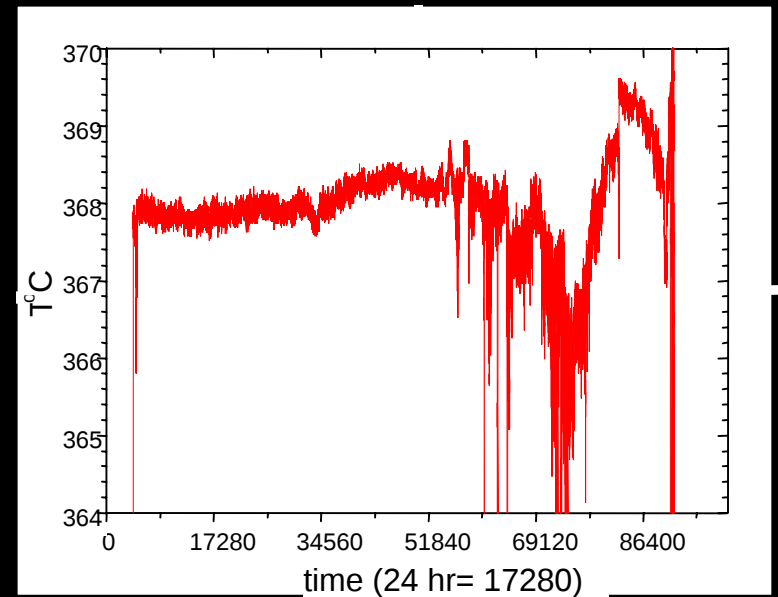
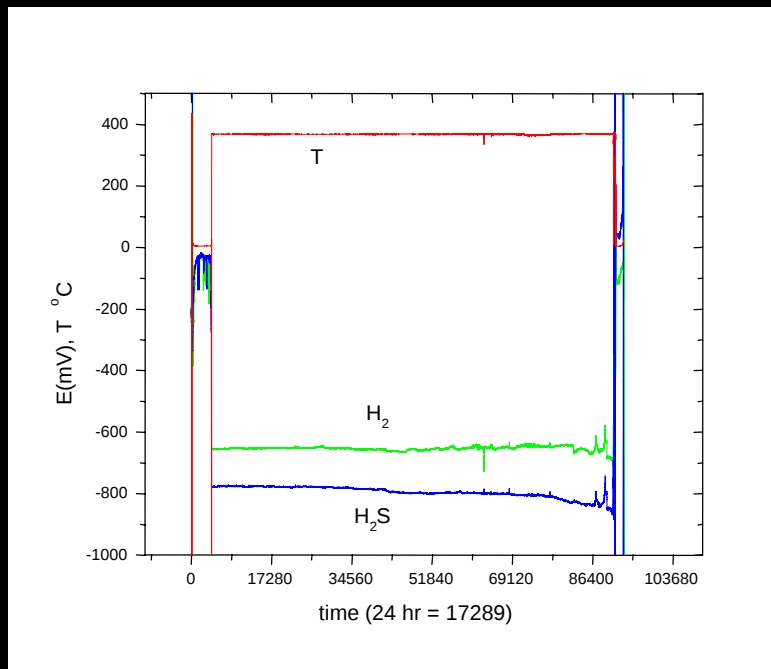
Theoretical Model of pH Change during Seawater-Vent fluid mixing



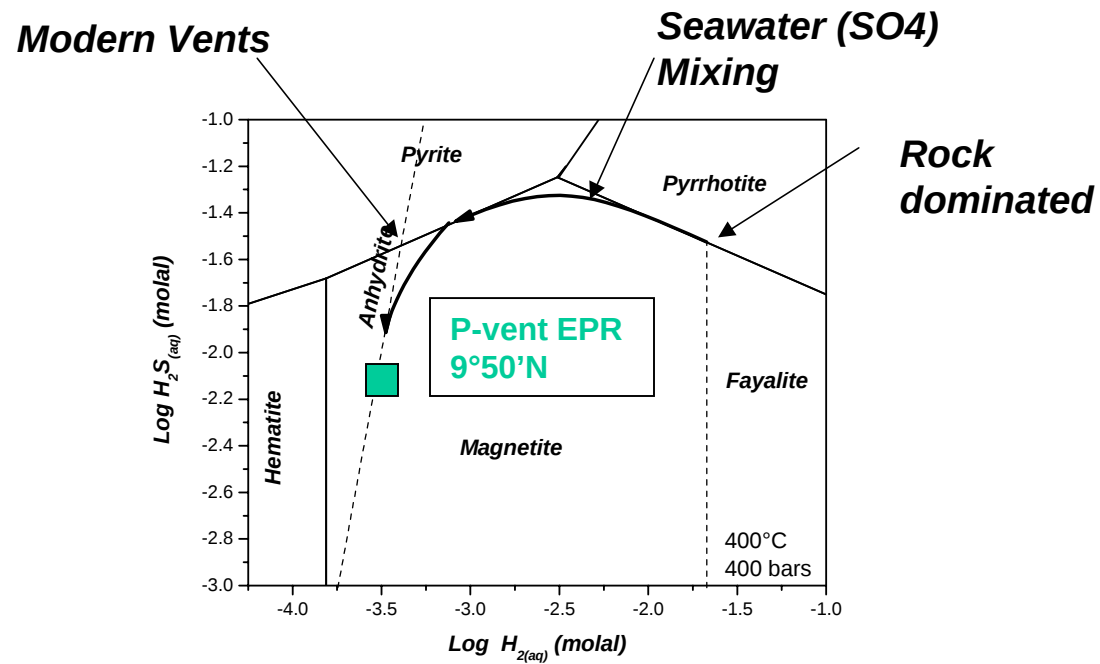
Sensor Deployment at P-Vent, EPR 9°50'N



Sensor Deployment at P-Vent, EPR 9°50'N: Temperature, H₂ and H₂S Record



Redox Relations/Evolution at Hydrothermal vents

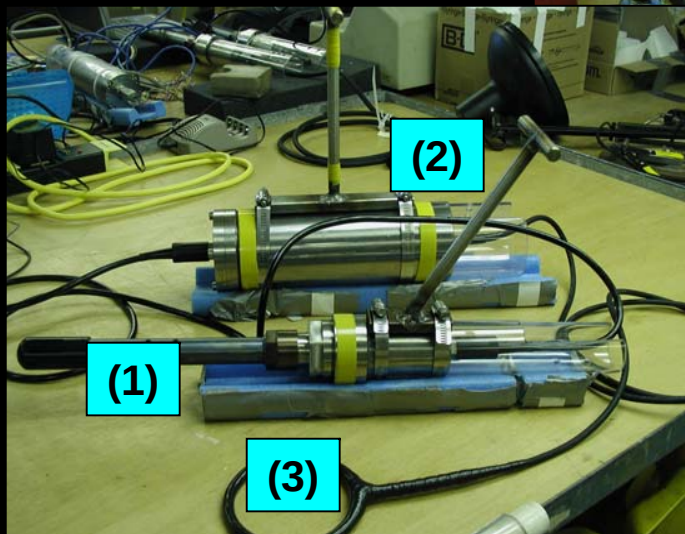
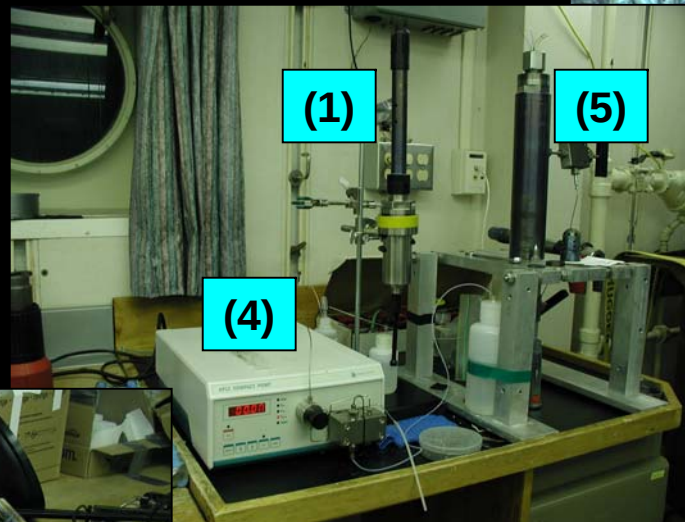
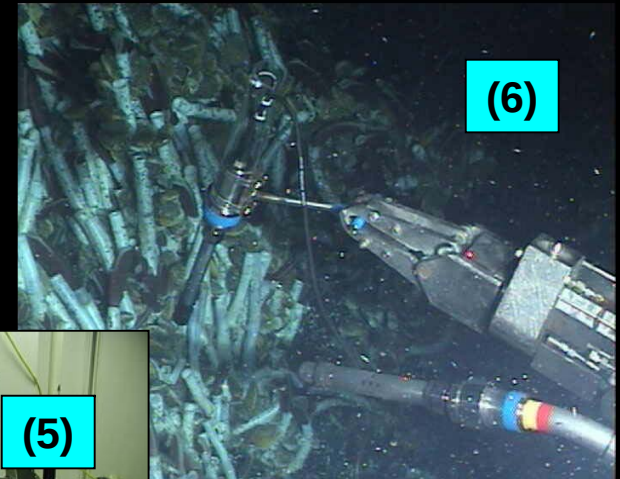


Tica Vent 9°50'N EPR

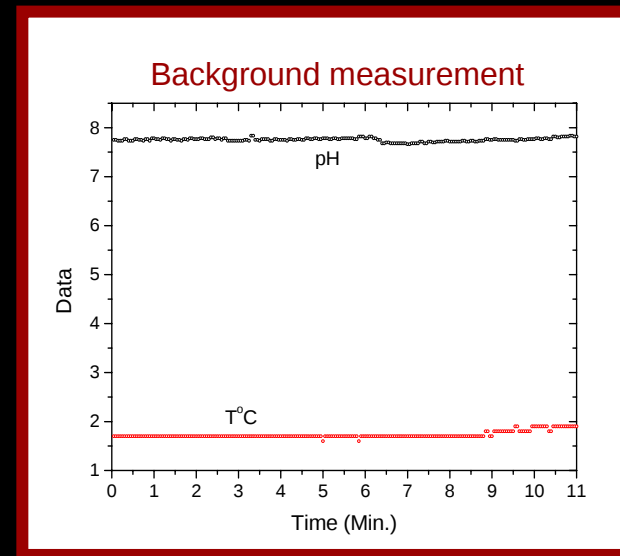
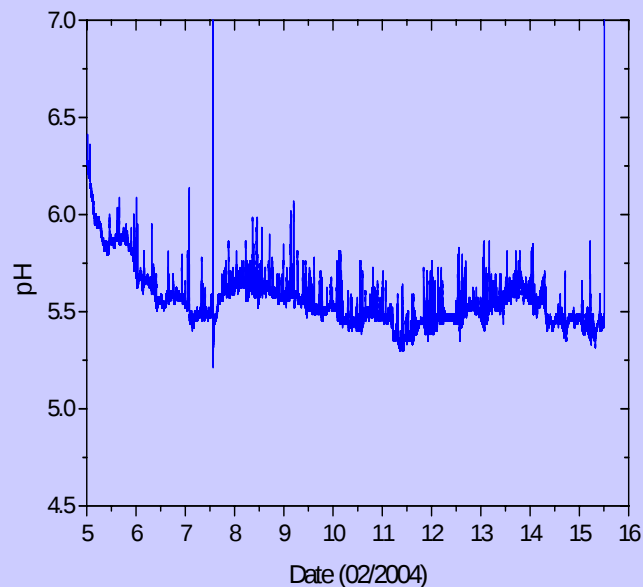
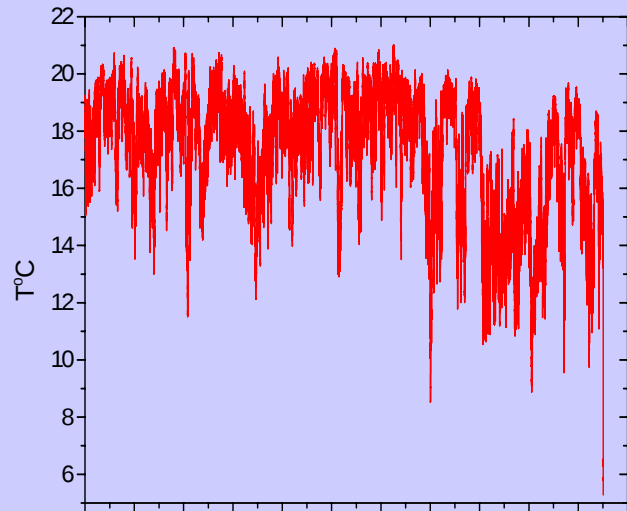


- Chemical and biological feedback
- Black smoker at ~ 330°
- Riftia community 8-26°C

The sensor unit was tested under the operational condition on ship board before and after deployment.



- (1) Sensor
- (2) Data logger
- (3) ICL communication line
- (4) HPLC pump
- (5) High Pressure testing vessel
- (6) Field deployment

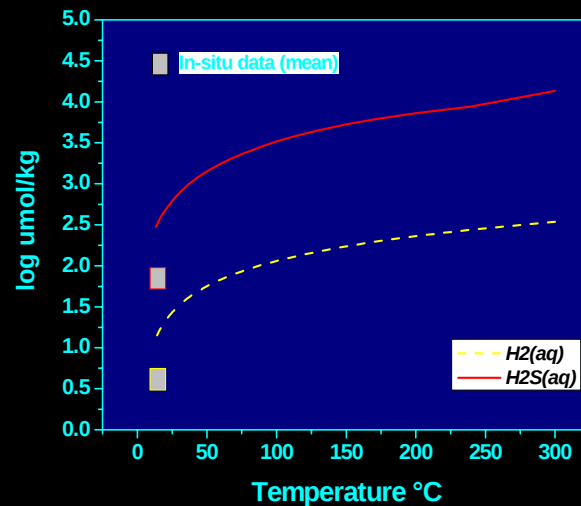


Temperature and pH(in-situ) at the base of tubeworm colony at Tica.

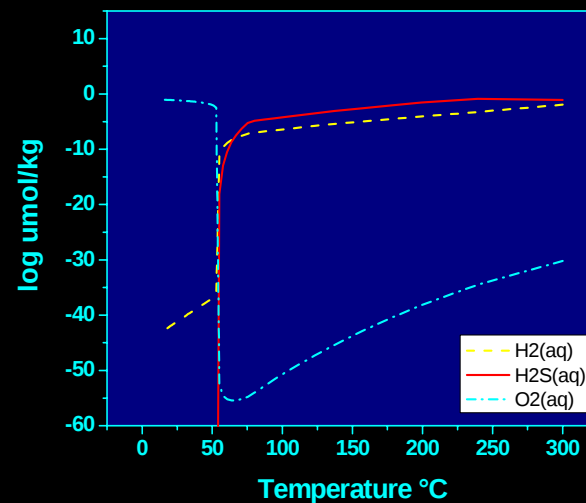
•Temperature variability--~12-22°C

•pH variability—6.5 to 5.5 with time-substantially lower than background seawater values.

In-situ H_2S and H_2 data in comparison with theoretical prediction



1. H_2 Redox decoupled during cooling



2. H_2 Redox coupled during cooling

Conclusions

- In-situ measurement and monitoring pH and Redox (H_2 , H_2S) of hydrothermal vent fluids has been demonstrated at a range of conditions and vent sites
 - High temperature vent fluids (EPR 9°N,P-vent)
 - Slightly acidic, only moderately reducing
 - » Compositional and temperature stability on the order of days
 - In-situ data confirm seawater mixing and pH lowering
 - Diffuse flow fluids at Tica reveal complex time series patterns involving temperature, pH, H_2 and H_2S
 - Seawater mixing and biogeochemical processes
 - Linkage between diffuse flow and high temperature venting?